



# Chemical Resistance and its Role in Resin Selection

Tuesday, November 1, 2022

Presented By: Cliff Watkins & Jeremy Bland

# Chemical Resistance and its Role in Resin Selection



- 40-year plastics industry veteran
- Past owner of custom compounder TP Composites
- PhD Chemistry
- 14 years with PPG Fiber Glass

**Cliff Watkins PhD**  
Direction, Application Development  
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- 23-year plastics industry veteran
- Pittsburg State University-Plastics
- Process engineering expertise
- Six Sigma Black Belt

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We are **THE FIXERS.**

# The PolySource Webinar Series



[HOW TO SURVIVE THE STRUCTURAL PA66 SHORTAGE](#)

[HOW TO IMPROVE YOUR MATERIAL SELECTION PROCESS](#)

[COMPARING NYLON-POLYAMIDES, INCLUDING SPECIALTIES](#)

[SELECTING THE RIGHT RESIN TO MEET YOUR APPLICATIONS REQUIREMENTS](#)

[RESOLVING PROCESSING AND PART PERFORMANCE ISSUES BY CHANGING MATERIALS](#)

[POLYKETONE \(POK\) – A MATERIAL OPTION VS EXPENSIVE AND SCARCE ENGINEERING THERMOPLASTICS](#)

[COMPARING RESIN MATERIAL TOUGHNESS AND HEAT RESISTANCE](#)

[MATERIAL OPTIONS VS SCARCE ENGINEERING THERMOPLASTICS](#)

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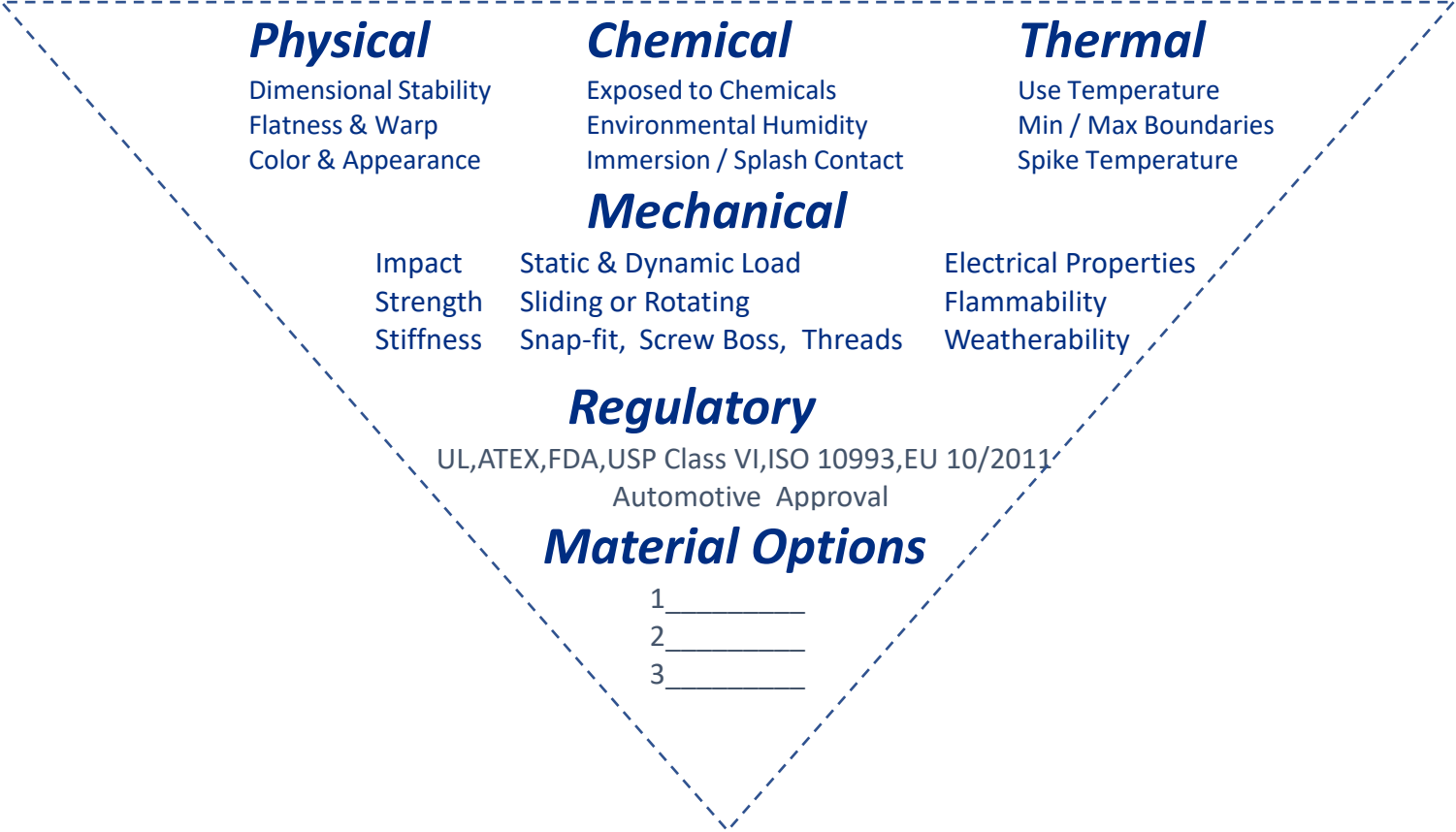
<https://polysource.net/resources/?type=webinar>

# Agenda

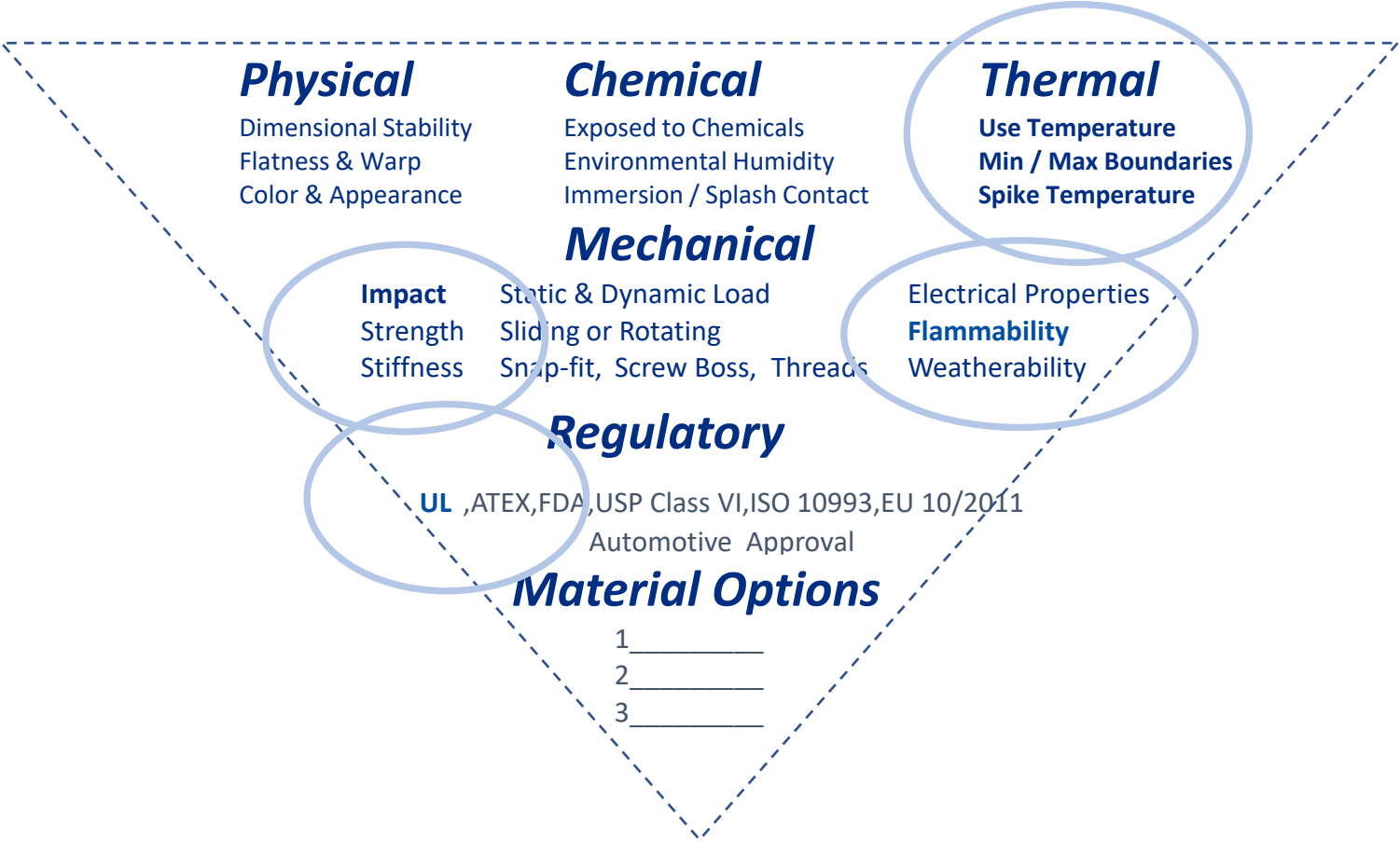


- Review of PolySource Design Funnel Methodology
- Chemical Resistance - Design/Material Selection/Processing/End-Use Considerations
- What is Chemical Resistance?
- Why is it Important?

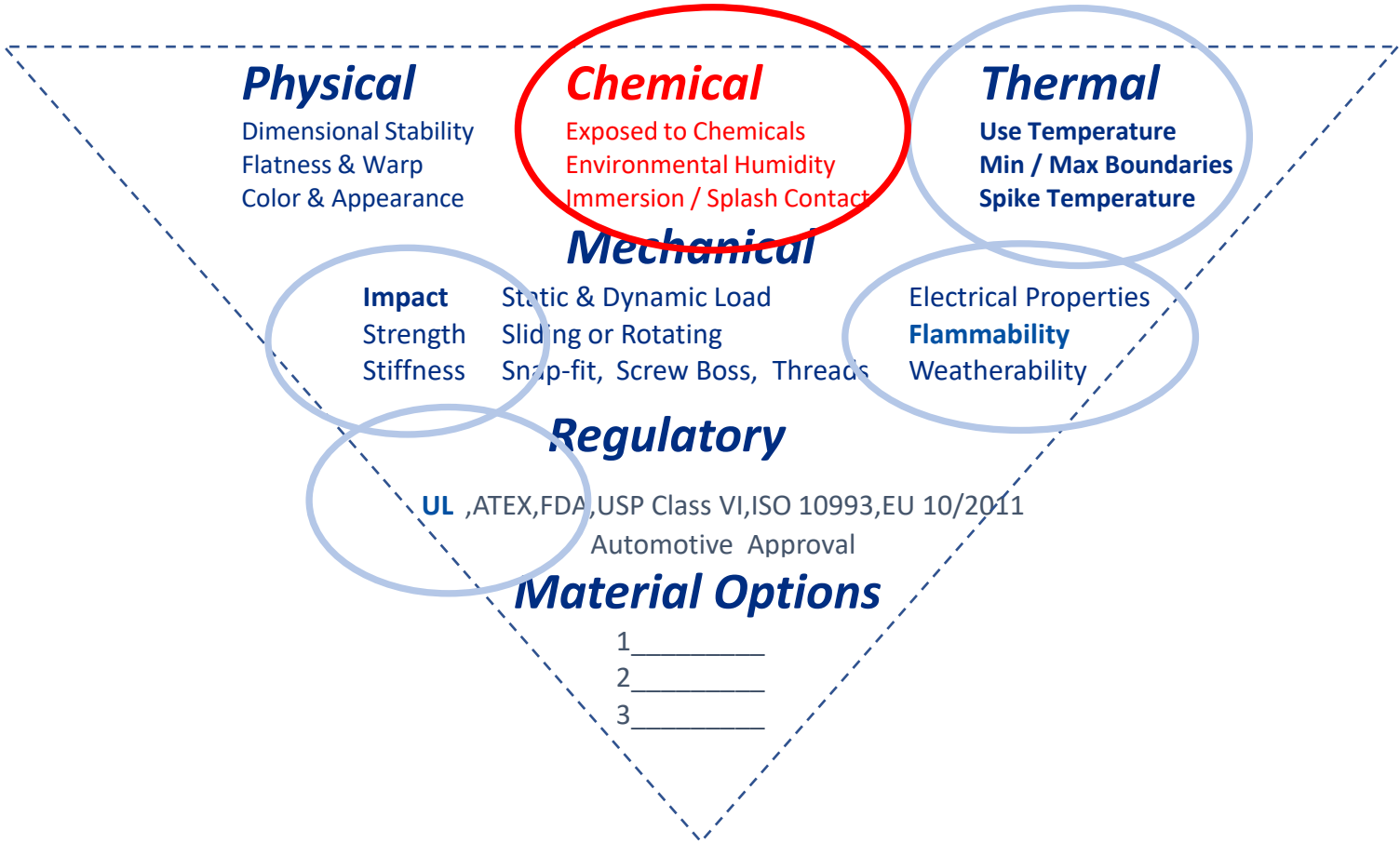
# The PolySource Design Funnel



# The PolySource Design Funnel



# The PolySource Design Funnel





## Chemical Resistance Test Methods

# Chemical Resistance and its Role in Resin Selection



## Chemical Resistance Test Methods

### Environmental Stress Crack Resistance (ESCR)

(ASTM D1693, ISO 22088-3) – exposure while strained – visual inspection for surface cracks and crazing



**Solvent Compatibility**

Solvent	ABS	LDPE	HDPE	PC	PMP	PP	PS	PTFE
Acetaldehyde	D	C	B	C	C	C	D	A
Acetic Anhydride	C	D	D	D	B	B	D	A
Acetone	D	C	C	D	A	A	D	A
Acid, Hydrofluoric	C	A	A	D	A	B	D	A
Acid, Trifluoroacetic	D	D	C	D	D	D	D	A
Acid, Acetic Dilute 50%	A	A	A	B	A	A	B	A
Acid, Hydrochloric 37%	C	A	A	D	B	D	C	A
Acid, Nitric	B	C	B	B	A	D	C	A
Acid, Sulfuric	D	B	A	C	B	C	C	A
Alcohol, Ethyl	A	B	A	B	B	B	B	A
Alcohol, Isobutyl	A	A	A	B	A	A	B	A
Alcohol, Methyl	D	A	A	B	A	A	C	A
Alcohol, n-Butyl	A	A	A	C	B	A	B	A
Alcohol, Propyl	B	A	A	D	-	A	A	A
Ammonium Hydroxide	B	B	A	-	B	B	B	A
Aniline	D	B	B	B	B	B	D	A
Aqua Regia	D	D	C	D	D	D	D	A
Benzaldehyde	B	B	A	C	B	A	D	A

**A** No Effect, excellent compatibility  
**B** Minor Effect, good compatibility  
**C** Moderate Effect, fair compatibility  
**D** Severe Effect, not recommended  
**-** No data available

Qualitative Visual Data

# Chemical Resistance and its Role in Resin Selection



## Chemical Resistance Test Methods

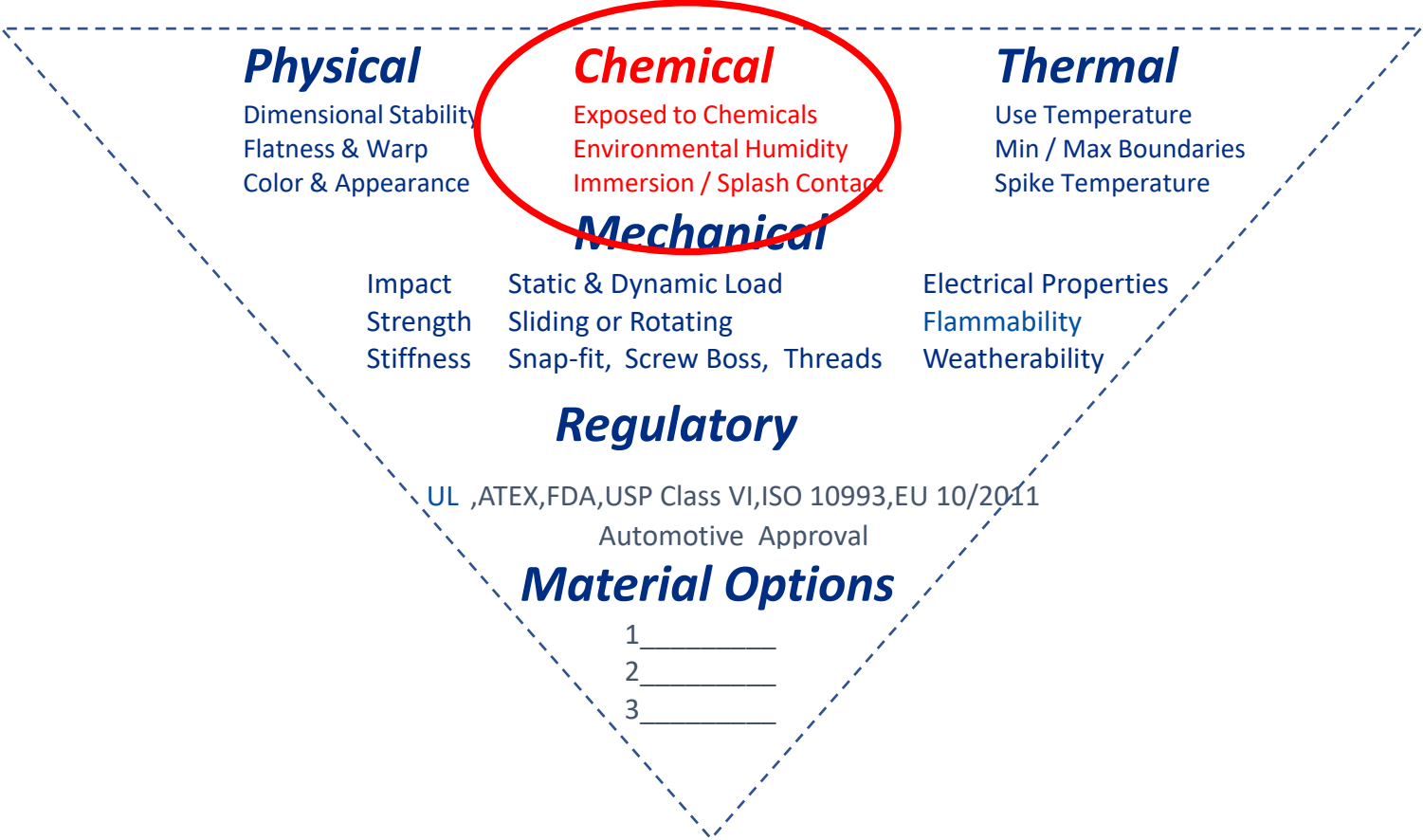
Immersion (ASTM D638, ISO 527, for example) – quantitative measurements for changes in mechanical & physical properties as a function of time, temperature and concentration. Soaking tensile bars in the chemical of interest, is most common, then measuring:



% Change in Strength, % Change in Elongation, % Change in Dimensions, Appearance

So, how do I find the Right Material?

# The PolySource Design Funnel



# Chemical Resistance and its Role in Resin Selection



Choosing the “Best” ETP for a specific chemical cannot be simplified down to choosing a semi-crystalline instead of an amorphous resin.

- What is the temperature environment?
- Is there a load on the part?
- Was the material molded correctly?
- Part thickness/dimensions/design?
- What is the concentration of the chemical?

*Don't Forget the Pink  
Elephant in the Room....  
HUMIDITY!*

Not Simply a Question of Semi-Crystalline vs Amorphous!

# Chemical Resistance and its Role in Resin Selection



Choosing the “Best” ETP for a specific chemical cannot be simplified down to choosing a semi-crystalline instead of an amorphous resin.

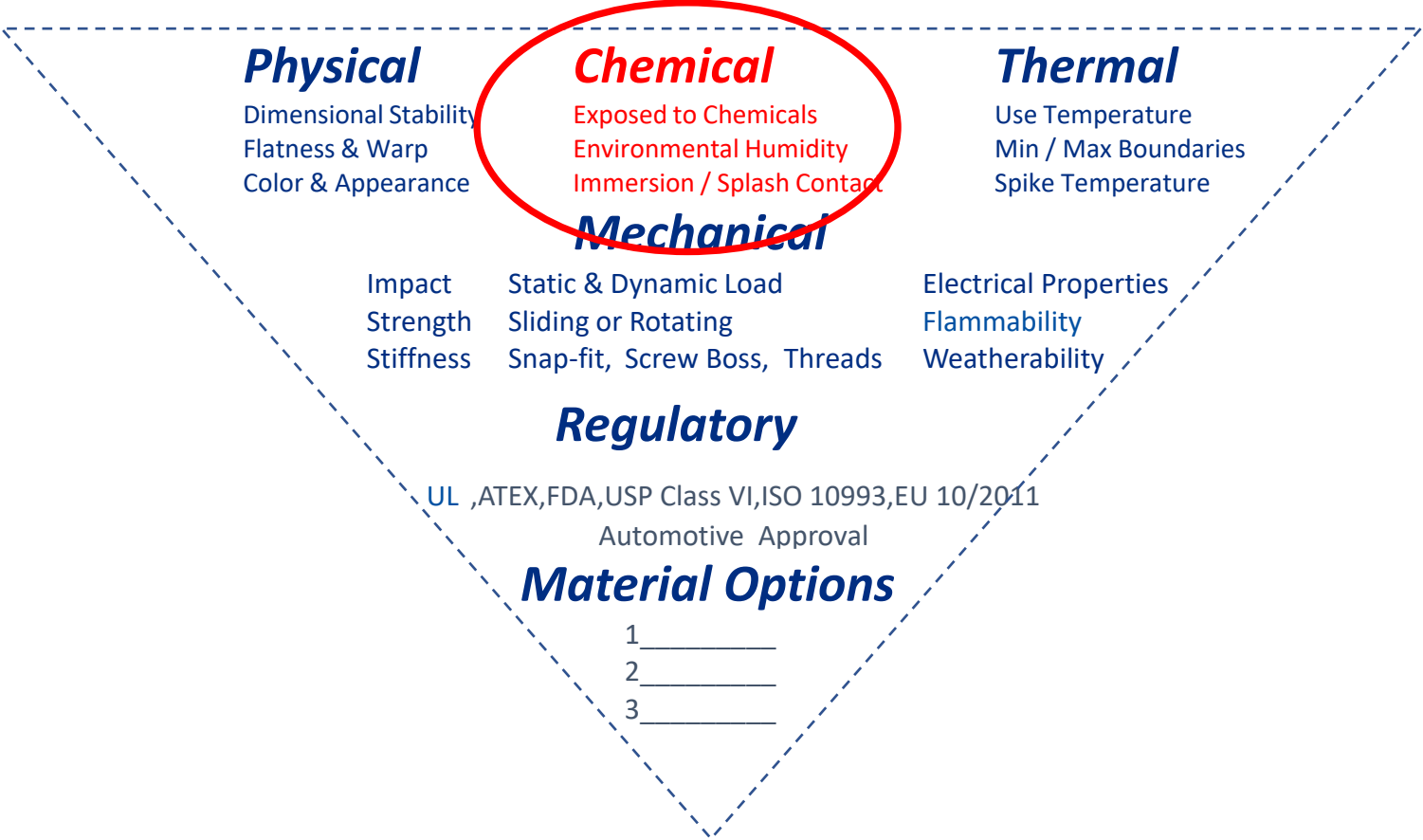
**Problem:** New application, customer selected PBT, because of this chart. They over-looked the use temp. at 90 °C in the design spec.

**Solution:** The PolySource solution: Use PPE/HIPS (XYRON from Asahi Kasei), because PBT degrades in hot/wet environments above 80 °C and XYRON is rated to 120 °C.

AGENTS, CONCENTRATION: WEIGHT-%	MATERIAL																	
	ABS	Acetal (copolymer)	Acetal (homopolymer)	Noryl®	Nylon 6	PBT	PEEK	PET	Polycarbonate	Polyethylene	Polypropylene (homopolymer)	PPS	PPSU (Radel® R)	PSU (polysulfone)	PTFE	PVDF	Ultem®	Vespel® SP-1
Transformer oil	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■
Trichlorethylene	▲	▲	▲	▲	●	▲	■	▲	▲	●	●	■	▲	■	■	■	■	■
Triethanolamine	■	■	▲	■	■	■	■	▲	■	■	●	■	■	■	■	■	■	■
Tylon B, aqueous solution 10%	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■
Urea, aqueous solution	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■
Vaseline	■	■	■	■	■	■	■	■	●	■	■	■	■	■	■	■	■	■
Water, cold	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■

Chemical Resistance – Amorphous Wins in Certain Applications

# The PolySource Design Funnel





# Chemical Resistance and its Role in Resin Selection



## The PolySource Design Funnel Questions

- What type of chemical?
- Is the chemical in a solvent?
  - Aqueous (waterborne)?
  - Organic?
  - Aerosol?
- What is the concentration?
- Duration of exposure?
  - Immersion?
    - Minutes/hours/days/weeks...?
  - Splash?
- Use temperature?
  - Constant versus cyclic?
- Applied stress?

Continuation of the use of .....”The PolySource Design Funnel”

# Chemical Resistance and its Role in Resin Selection



## Some Examples of Chemicals Causing Problems

### Bio Fuels – POM to PPA

The issue: Canola and Rapeseed oils: erucic acid “C22 fatty acid”

*Operating temperatures and the inherent acidity forced the conversion away from POM to PPA-based materials \$\$\$*

### Mouthwash Cap – Not Solved Yet

The issue: ethanol, quaternary ammonium compounds or “quats”, benzoic acid, etc contained in mouthwash attack the plastic causing leaks. The ubiquitous foil layer under the cap prevents attack and ensures a tight leak-proof seal

*Trying to consolidate the design to eliminate the foil inner cap liner is proving a real challenge*

Unexpected Chemical Interactions Created Lots of Extra Work

# Chemical Resistance and its Role in Resin Selection



## Some Examples of Chemicals Causing Problems

### Lithium grease – PC to PC/PBT Blend

The issue: The mineral oil carrier used in the lithium grease had enough aromatic impurities to attack the PC

*Adding grease to fix a 'squeak' in a gear train led to an 18-month delay*

### Synthetic whip cream - PC to PC/PBT Blend

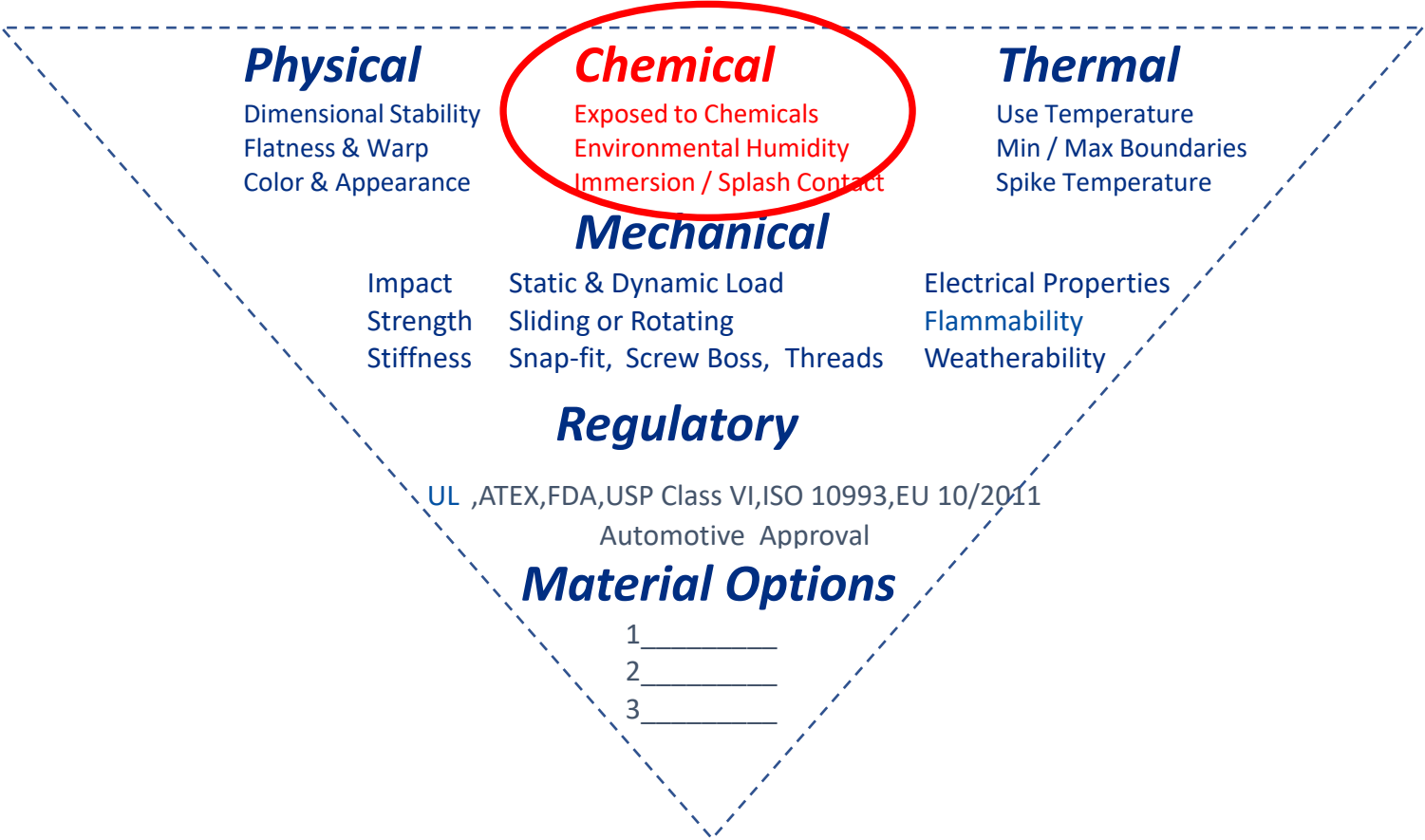
The issue: Unsaturated fatty esters are used in Cool Whip-type synthetic whipped cream and these additives caused stress cracks and crazing on commercial beverage bezels. The contact occurred as a result of the barista using the same sponge or cloth to mop up spills and then wiping down the machine.

**\*\*Shout out to Jeff Jansen at The Madison Group for the very elegant lab analyses\*\***

*The surge in the demand of specialty coffee caused a major surface crazing issue and re-design*

Unexpected Chemical Interactions Created Lots of Extra Work

# The PolySource Design Funnel



# Today's Webinar



The screenshot shows the PolySource website homepage. At the top left is the PolySource logo with the tagline "Resin • Solutions • Delivered". To the right of the logo are five blue buttons: "816.540.5300", "LINE CARD", "DATA SHEETS", "GLOBAL SUPPLY", and "REQUEST A QUOTE". Below these is a horizontal navigation menu with links for "HOME", "PRODUCTS", "RESINS", "SUPPLIERS", "RESOURCES", "MARKETS", "ABOUT US", and "CONTACT US". A search icon is on the far right. A blue dropdown menu is open under "RESOURCES", with "WEBINARS" highlighted. The main content area features a large image of blue resin granules on the left and three men in suits on the right. Text on the left reads "EXPLORE A BETTER RESIN-BUYING EXPERIENCE" with a "FIND OUT HOW" button below it. At the bottom right, a yellow banner says "We are THE FIXERS" with "REQUEST A QUOTE" and "HAVE A PROBLEM?" buttons below it.

Please Find the Presentation @[www.polysource.net/resources/webinars](http://www.polysource.net/resources/webinars)

## Thank You for Joining the Discussion Today!!



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- Past owner of TP Composites-  
bought by Techmer PM in 2013
- PhD Chemistry
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QUESTIONS????